

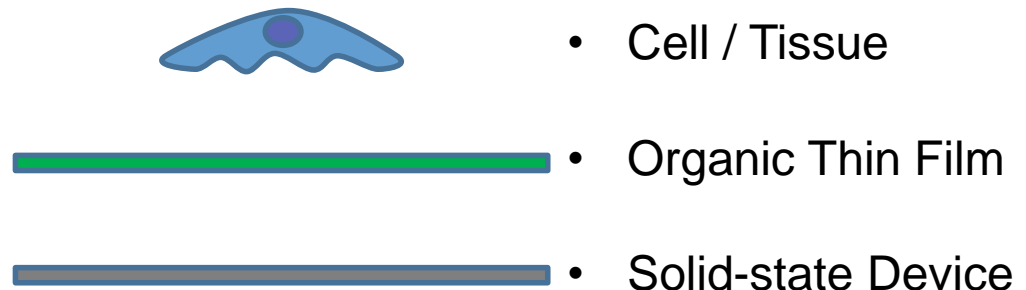
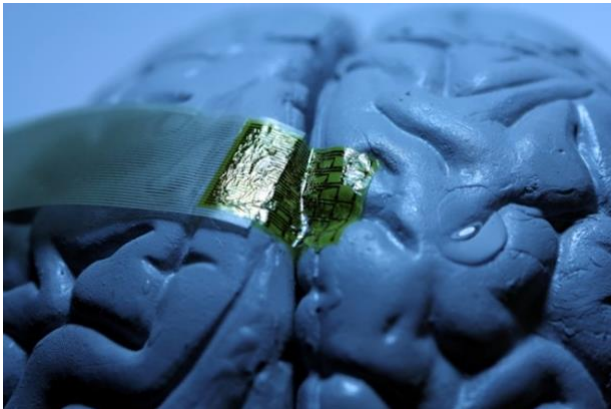
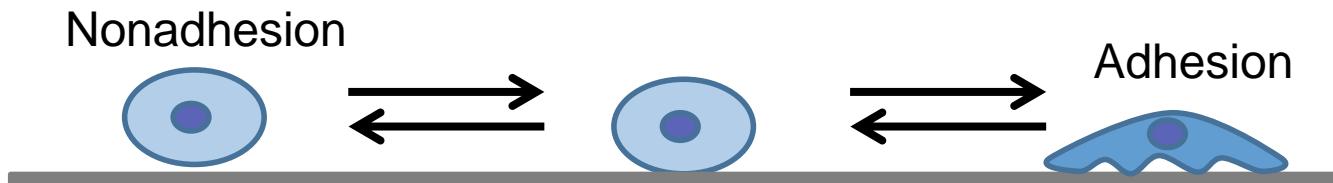
MOLECULAR VAPOR DEPOSITION AND PATTERNING OF ORGANOSILANE SELF-ASSEMBLED MONOLAYERS FOR DIRECTED GROWTH OF NEURON CELLS.

Felix Alfonso & Hsin-Ya Lou
EE412 Final Presentation
Mentor: Dr. Michelle Rincon
Dr. J Provine



SNF Stanford
Nanofabrication
Facility

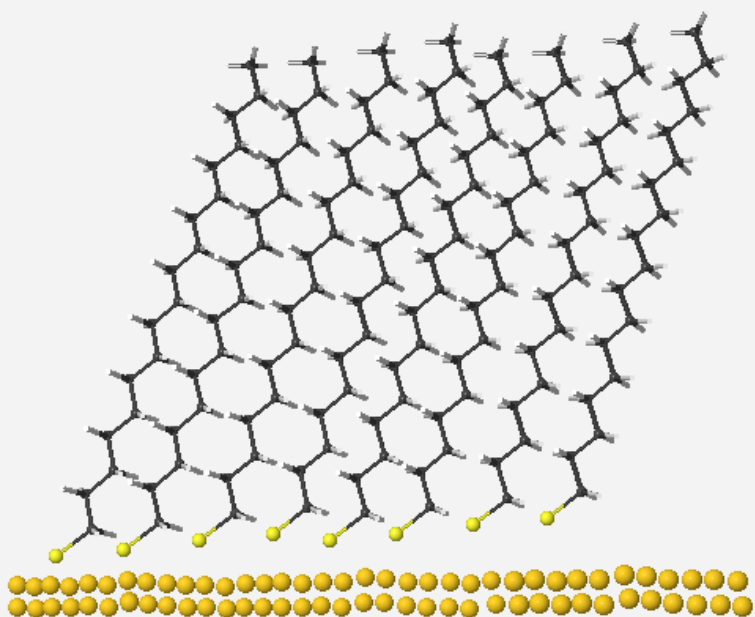
Solid State Devices and Biotechnology



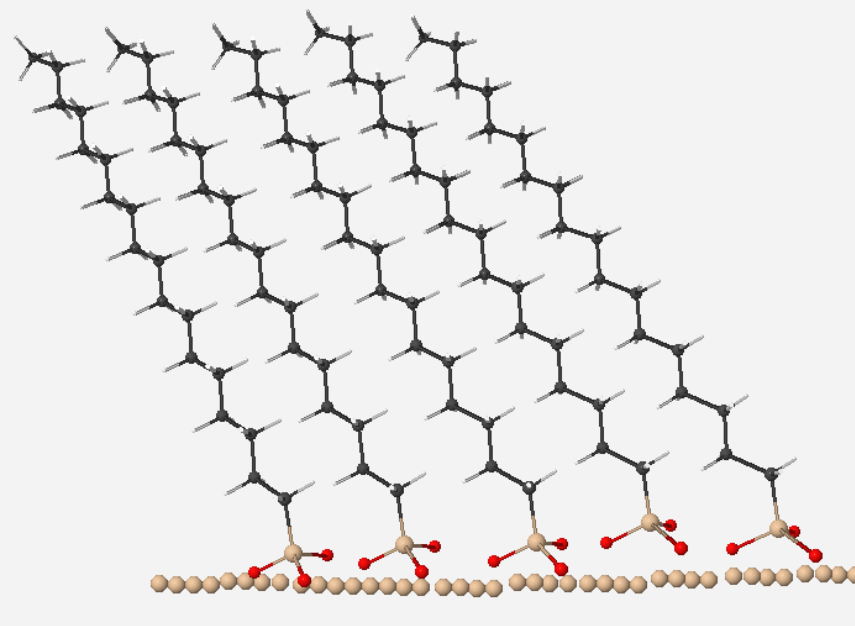
- In order to facilitate the integration of solid state devices and biological materials extensive focus has been given to organic thin films as a mediator.

Self Assemble Monolayers (SAMs)

- Alkanethiol

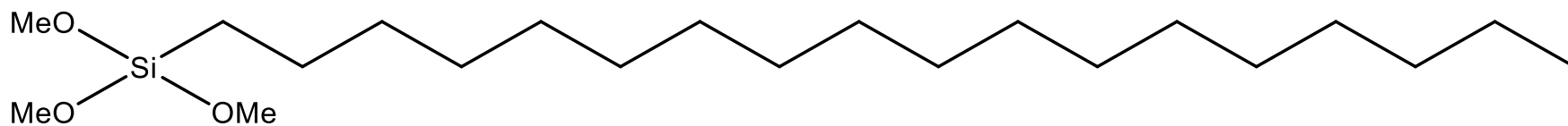


- Organosilane

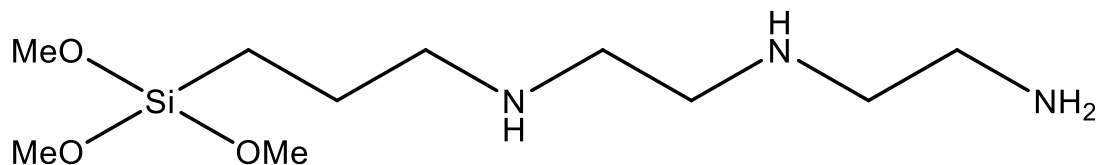


- Alkanethiol monolayers on planar Au surfaces undergo oxidation upon prolonged exposure to air.
- Organosilane mechanical and chemical stability in ambient environment once they have been anchored to a surface.

Organosilane used in this project



- Octadecyltrimethoxysilane (ODTS)
 - Hydrophobic, cytophobic



- Diethylenetriaminetrimethoxysilane (DETA)
 - Hydrophilic, cytophilic

ODS-DETA deposition and Patterning

A

Oxygen Plasma Clean
Drytek2

B

Deposition of ODS
MVD

C

Addition of Resist
svgcoat

D

Pattern By lithography
Karlsuss

E

Dissolved exposed area with
Organic solvent / Oxygen Plasma

F

Deposition of DETA
MVD

G

Remove Resist
Lift-off

— Silicon dioxide

— ODS

— DETA

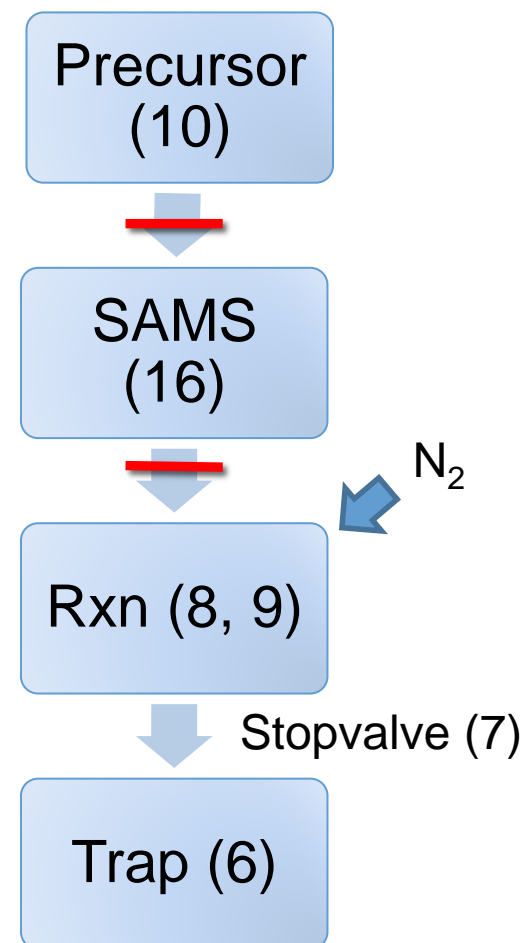
— Photoresist

Deposition recipe

1. Preheat:

1	Flow (N ₂)		20	sccm
2	Heater	6 (Trap/Pump line)	130	Deg C
3	Heater	7 (Stopvalve)	150	Deg C
4	Heater	8 (Reaction Chamber)	150	Deg C
5	Heater	9 (Reaction Chamber)	150	Deg C
6	Heater	10 (Precursor)	120	Deg C
7	Heater	16 (SAMS Chamber)	135	Deg C
8	Stabilize	6		
9	Stabilize	7		
10	Stabilize	8		
11	Stabilize	9		
12	Stabilize	10		
13	Stabilize	16		
14	Wait		600	sec

Direction of gas flow



Deposition recipe

2. Chamber Purging

15	Stopvalve		0 (Close)	
16	Wait		60	sec
17	Stopvalve		1 (Open)	
18	Wait		60	sec
19	Goto	16	3	

3. Fill Precursor

20	flow		0	sccm
21	Wait		20	sec
22	SAMS Fill	0	1	torr

Direction of gas flow

Precursor
(10)



SAMS
(16)



Rxn (8, 9)



Stopvalve (7)

Trap (6)

Deposition recipe

4. Reaction

22	Stopvalve		0	
23	Wait		0.5	sec
24	Pulse	0	30	sec
25	Wait		3600	sec
26	Stopvalve		1	

5. Remove Precursor

27	Wait	240		sec
28	flow		20	sccm

Direction of gas flow

Precursor
(10)



SAMS
(16)



Rxn (8, 9)

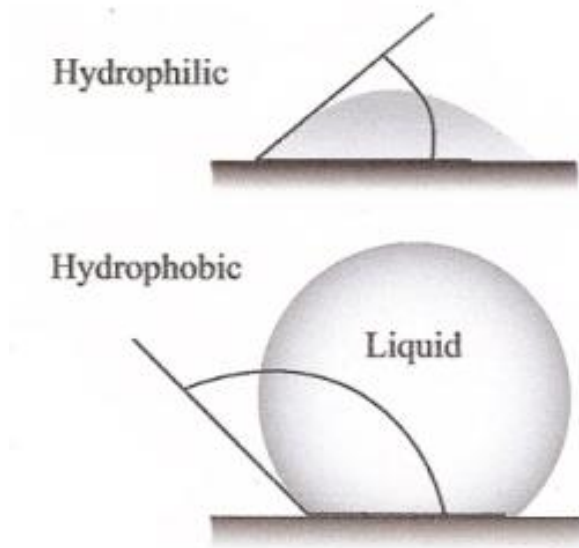


Stopvalve (7)

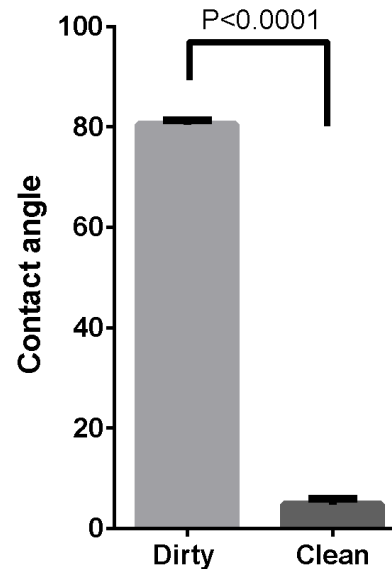
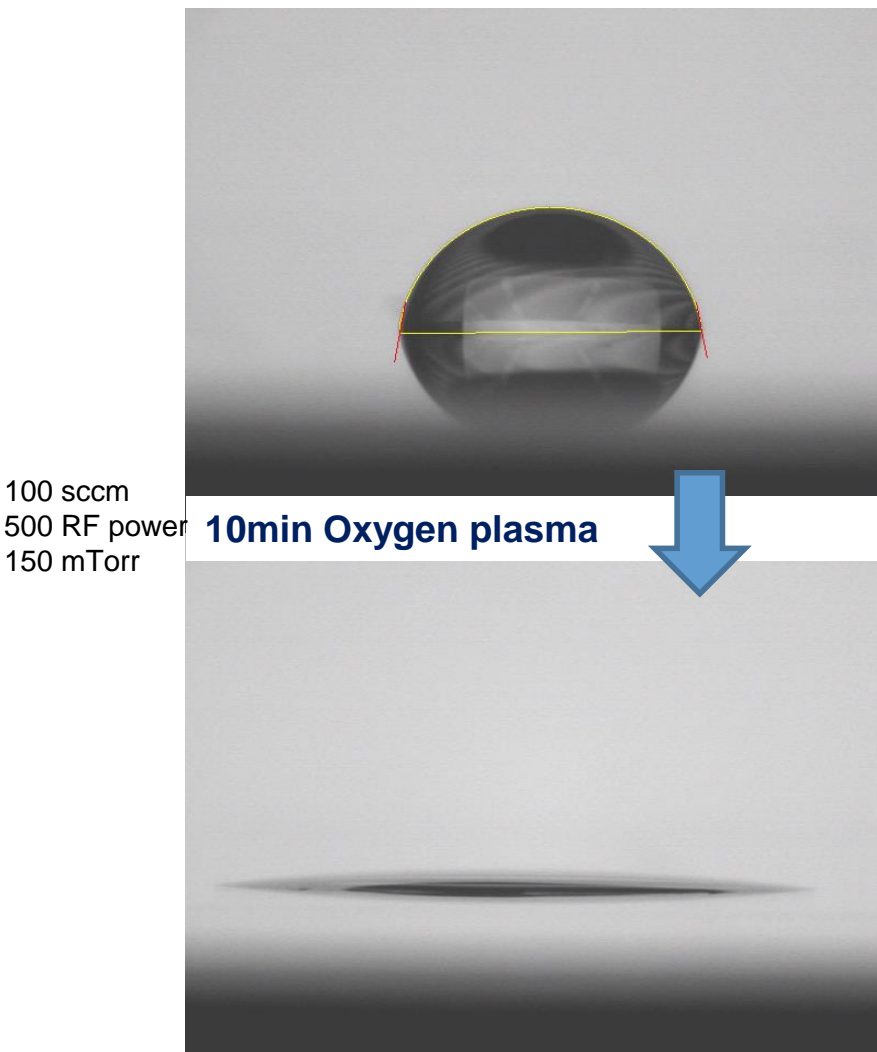
Trap (6)

How to quantify the deposition efficiency?

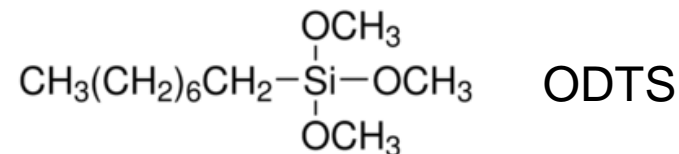
- Contact angle measurement
- Visualize the hydrophobicity of the surface
- Visualize the uniformity by multi-place measurement



Contact angle and surface cleanliness



- Wafers with cleaner (or hydrophilic) surface have smaller contact angle.



- After ODTS deposition, the surface should become hydrophobic, which results in higher contact angle.

Deposition on different surface



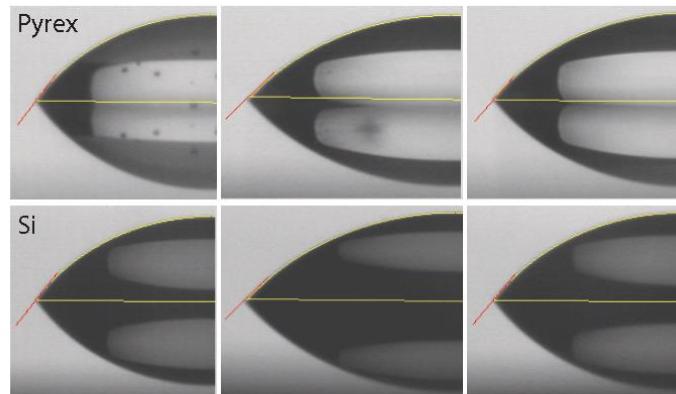
Silicon



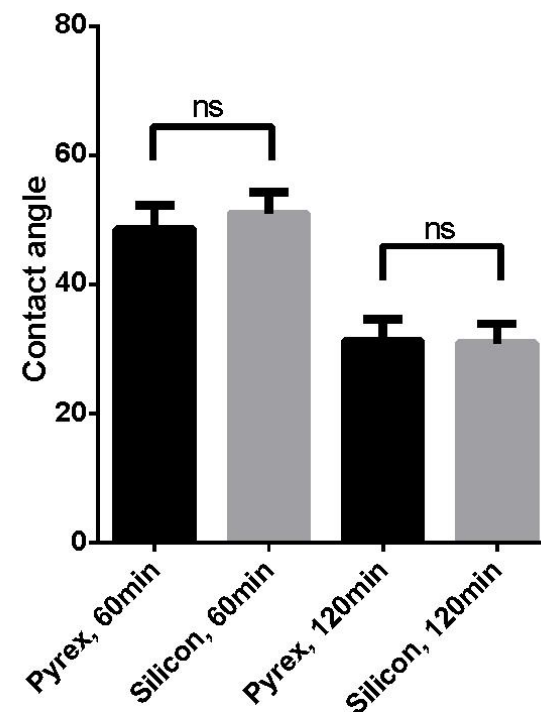
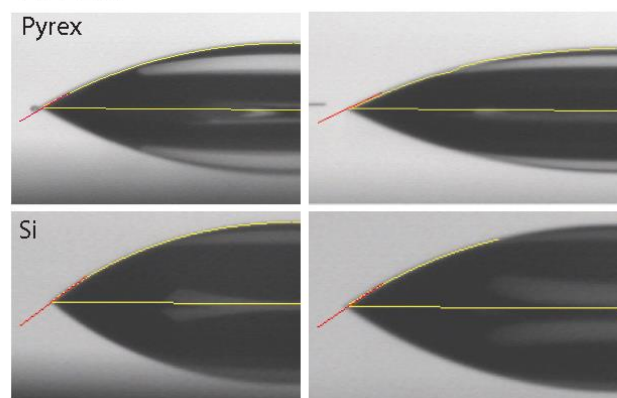
Pyrex®

(Contain O, Si,
Na, P, B, and K)

ODS 1hr



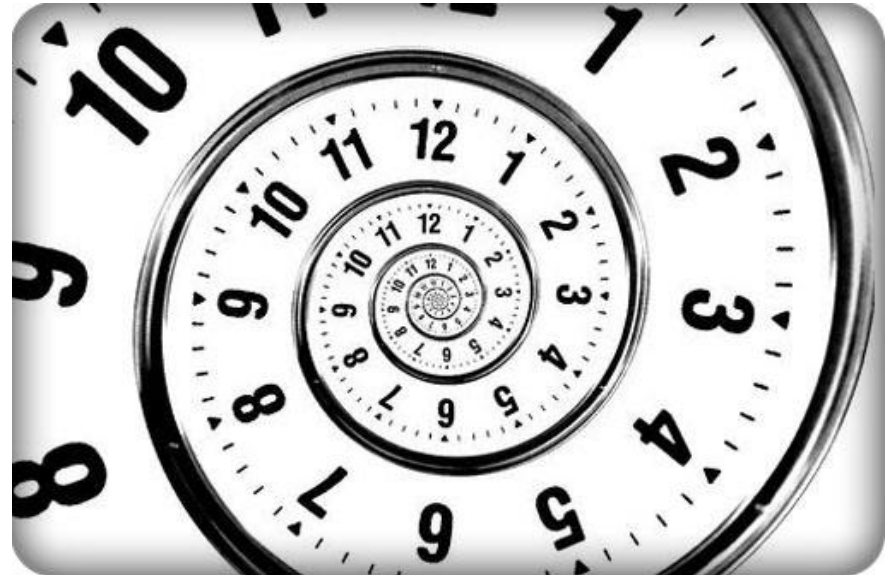
ODS 2hr



- ODS deposition characteristic for both wafers is similar.

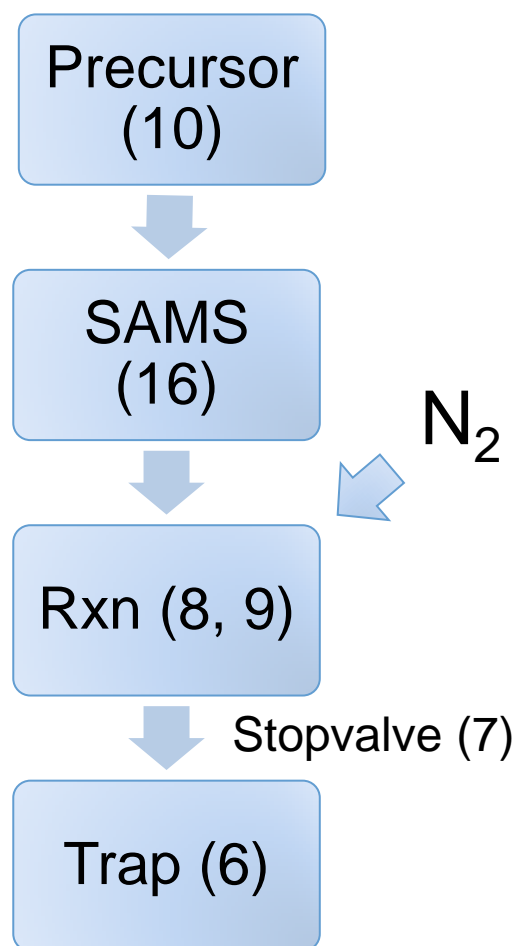
Disadvantages

- Inefficiency of refilling precursor chamber
 - About 0.05~0.1 torr/hr, need about 6 hours to refill the SAMs chamber

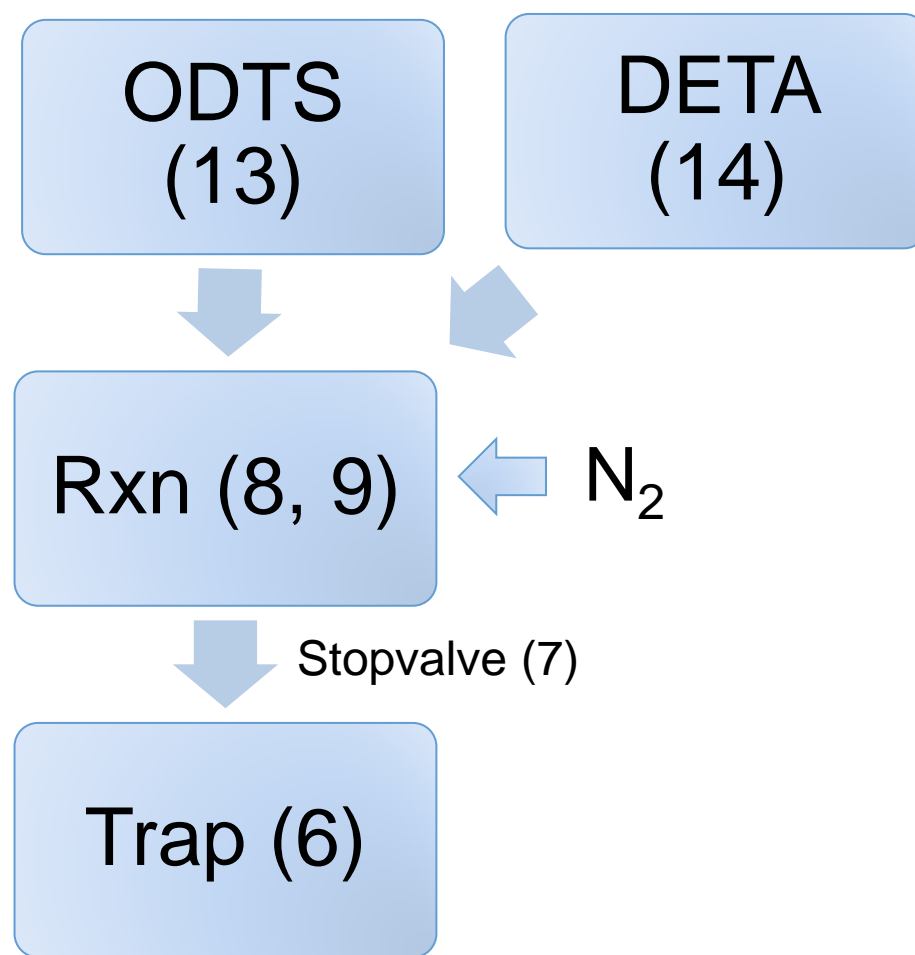


Modify the recipe

Old version

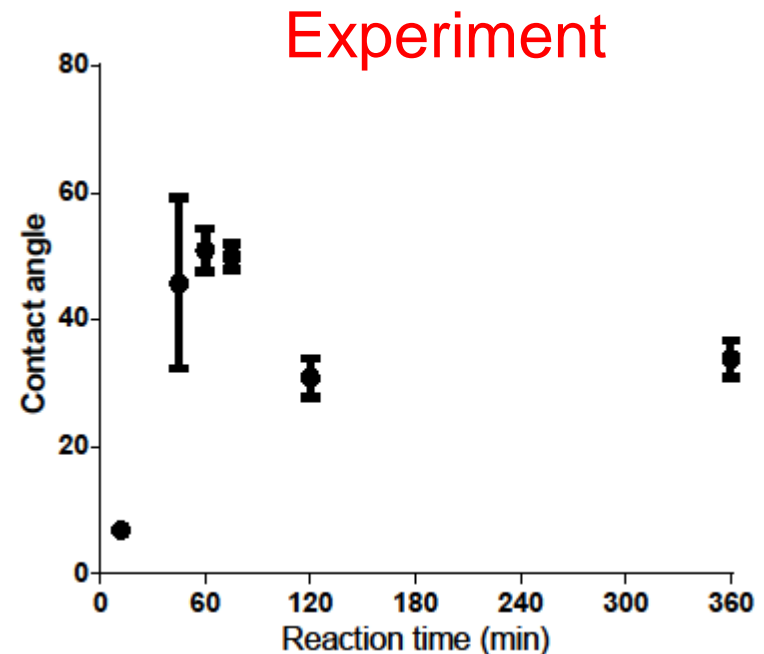
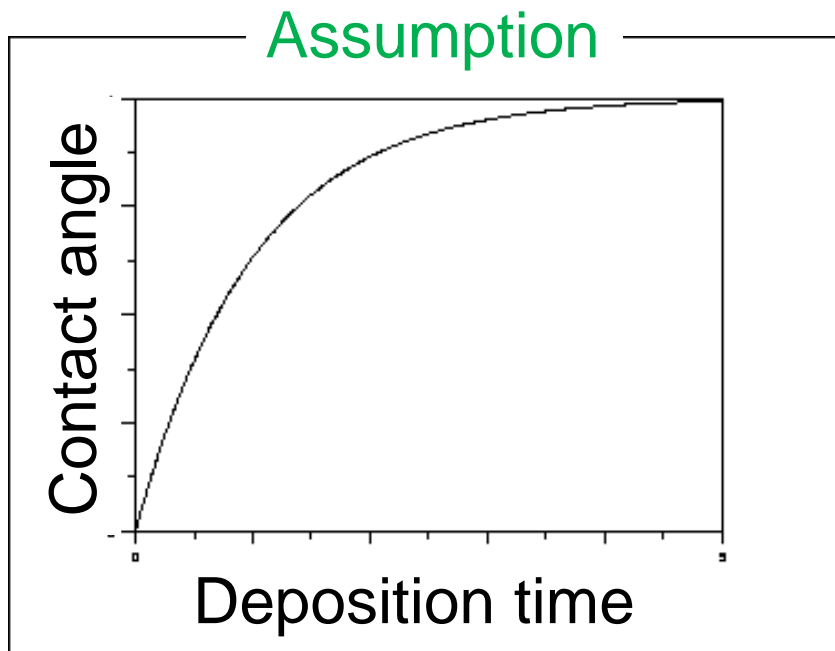


New version



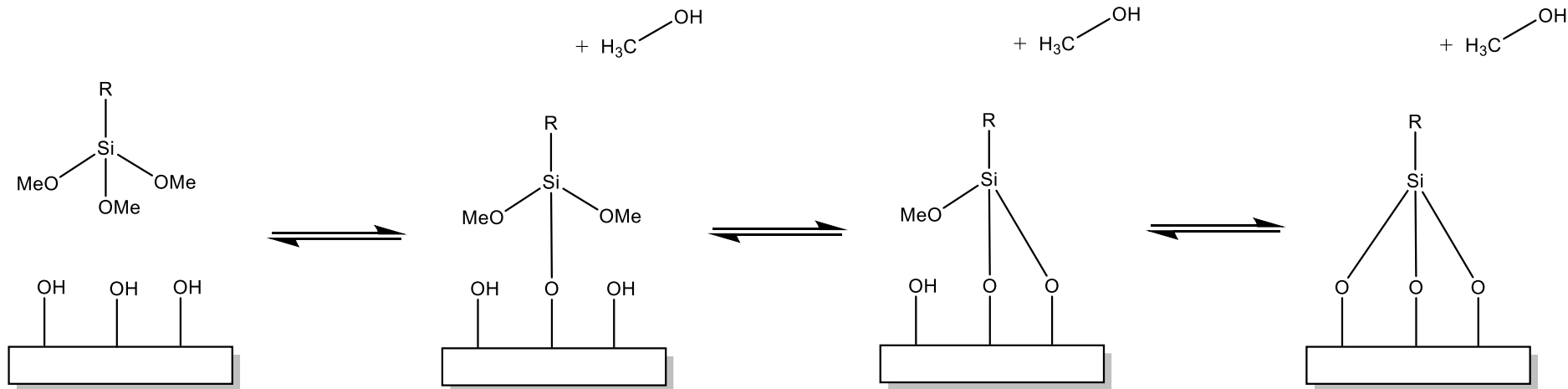
ODS deposition

- Assumption: Longer deposition time, higher contact angle.



- Experiment: Has local maxima, and reach to equilibrium.

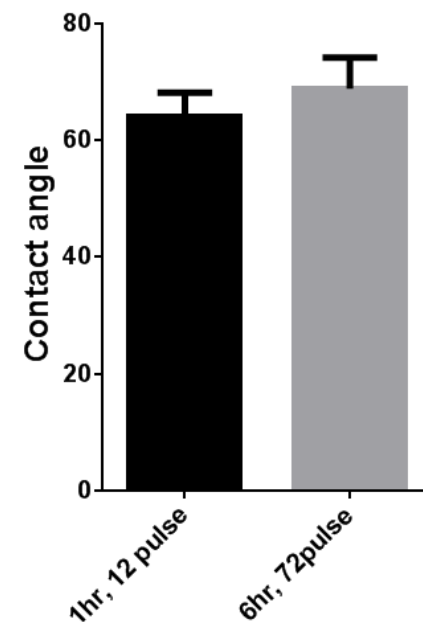
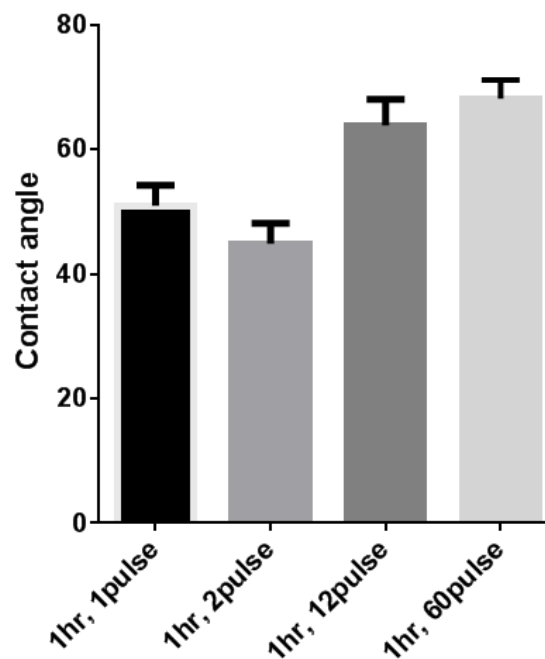
Mechanism of ODTs deposition



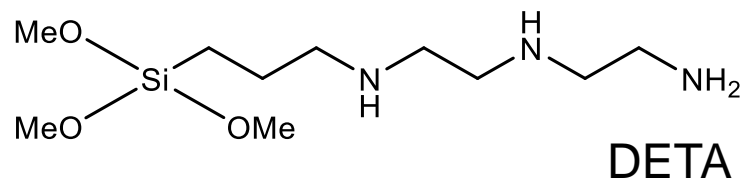
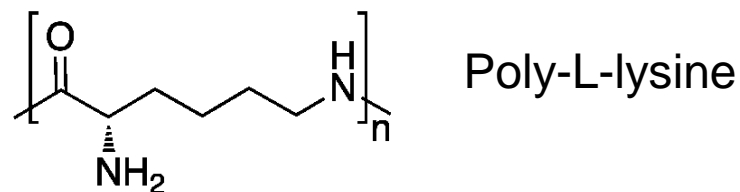
- When ODTs reacts with hydroxyl groups on the surface, methanol will appear as side product.
- Since the reaction is reversible, the more the methanol, the faster the reverse reaction.
- Once the reverse reaction is faster than forward reaction, the deposition efficiency will decrease.

Proof the mechanism

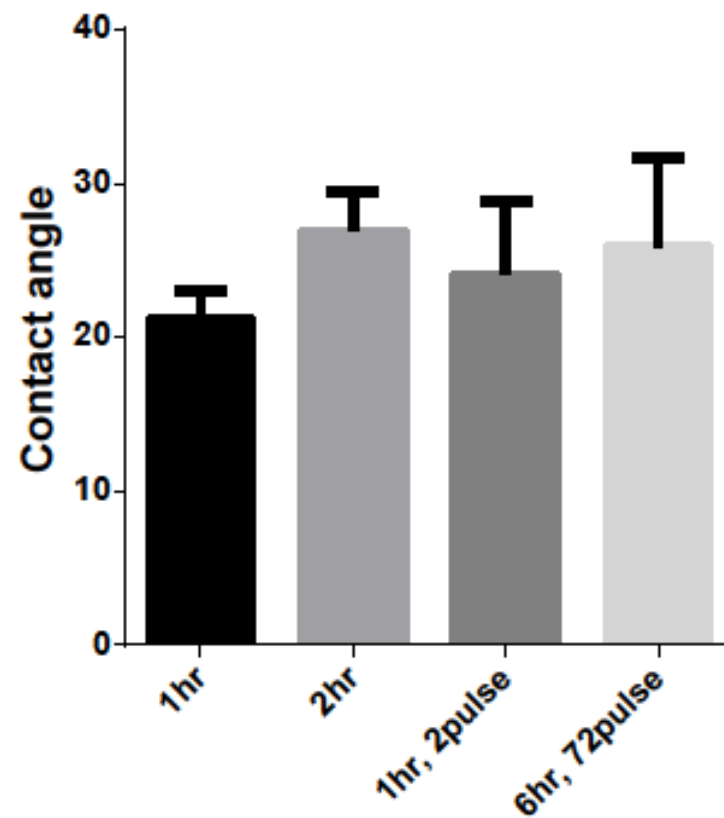
- **Multiple pulses**
- Before every pulse, methanol will be removed and new ODTS vapor will enter into the chamber.
- Results showed that increasing pulses will increase contact angle.



DETA deposition

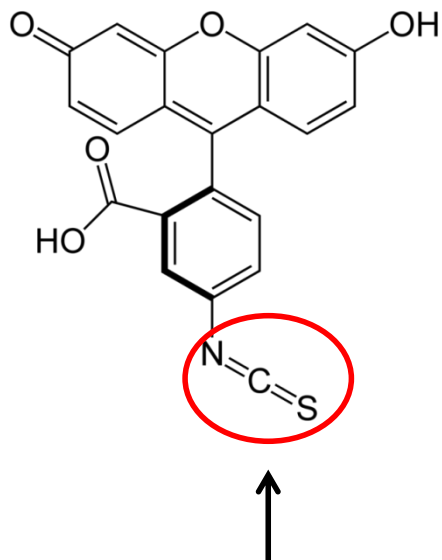


- DETA could promote cell adhesion as poly-L-lysine
- Hard to monitor the deposition efficiency by contact angle.

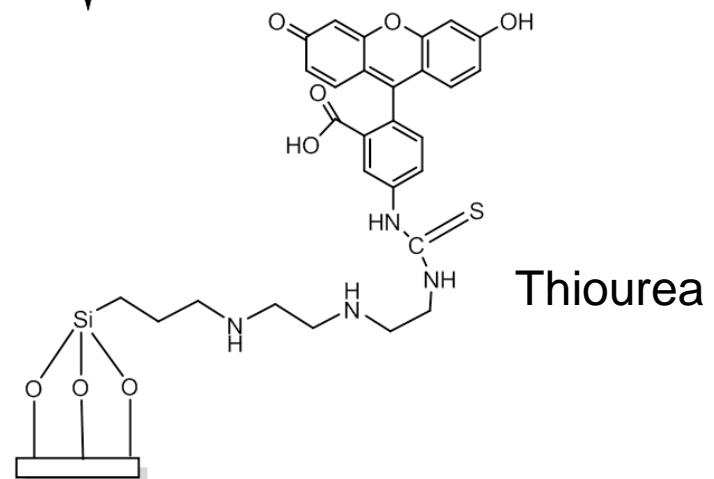
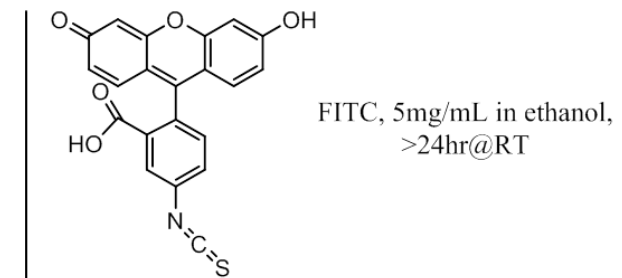
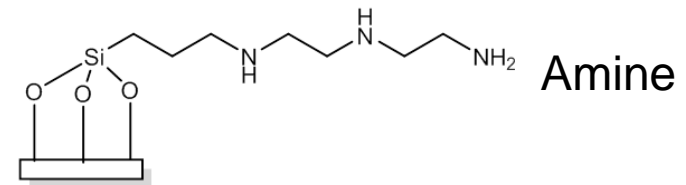


Characterization of DETA deposition

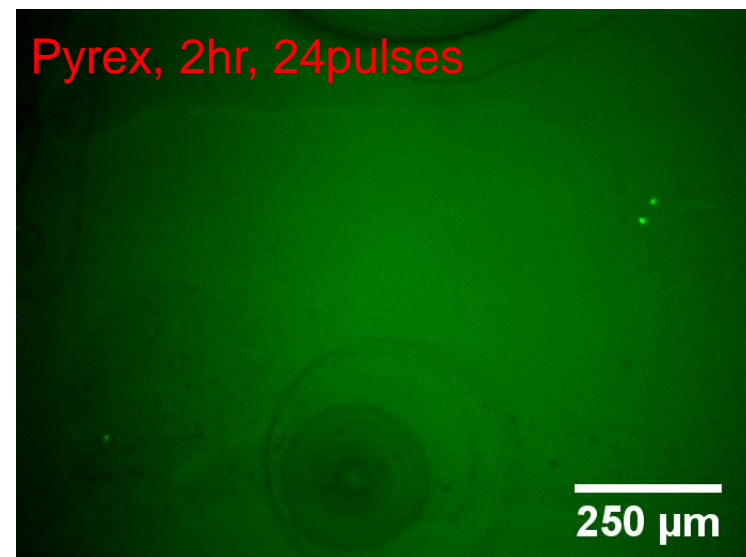
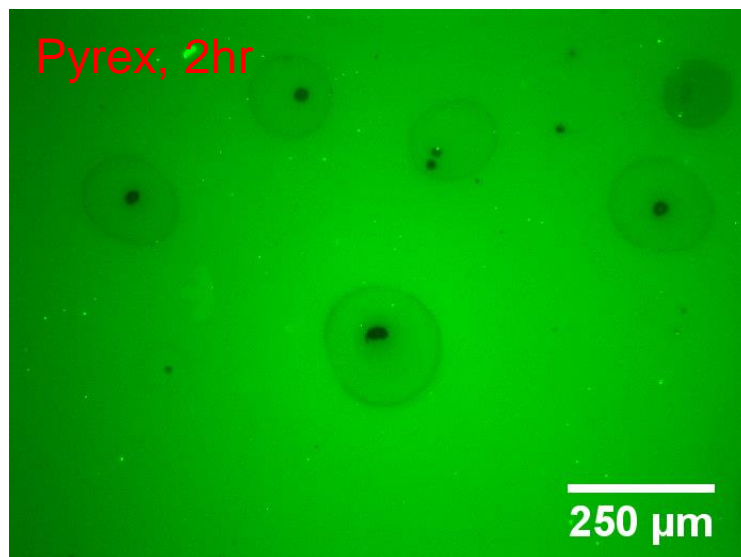
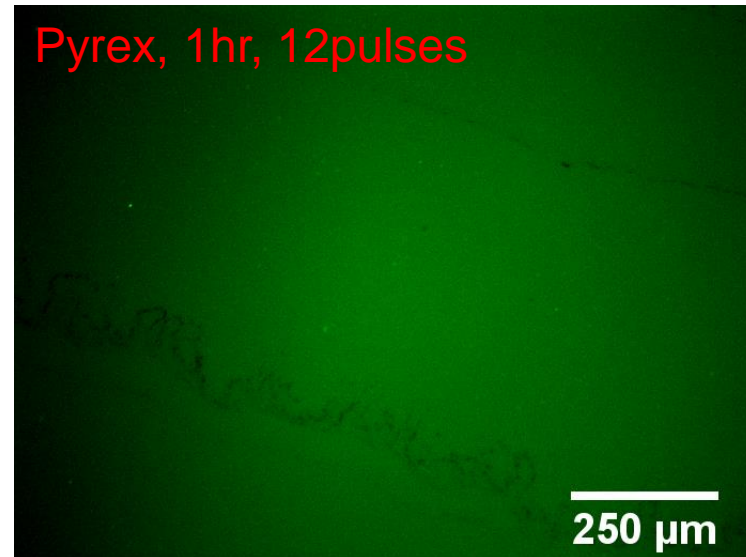
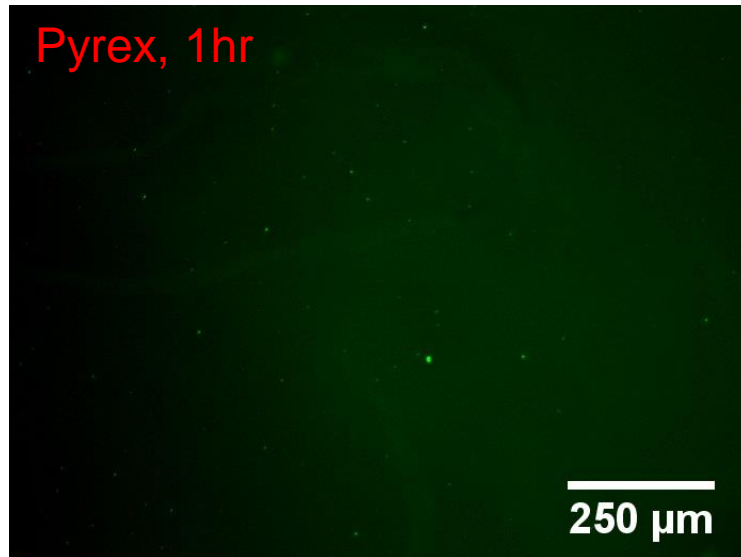
- Fluorescein isothiocyanate (FITC)



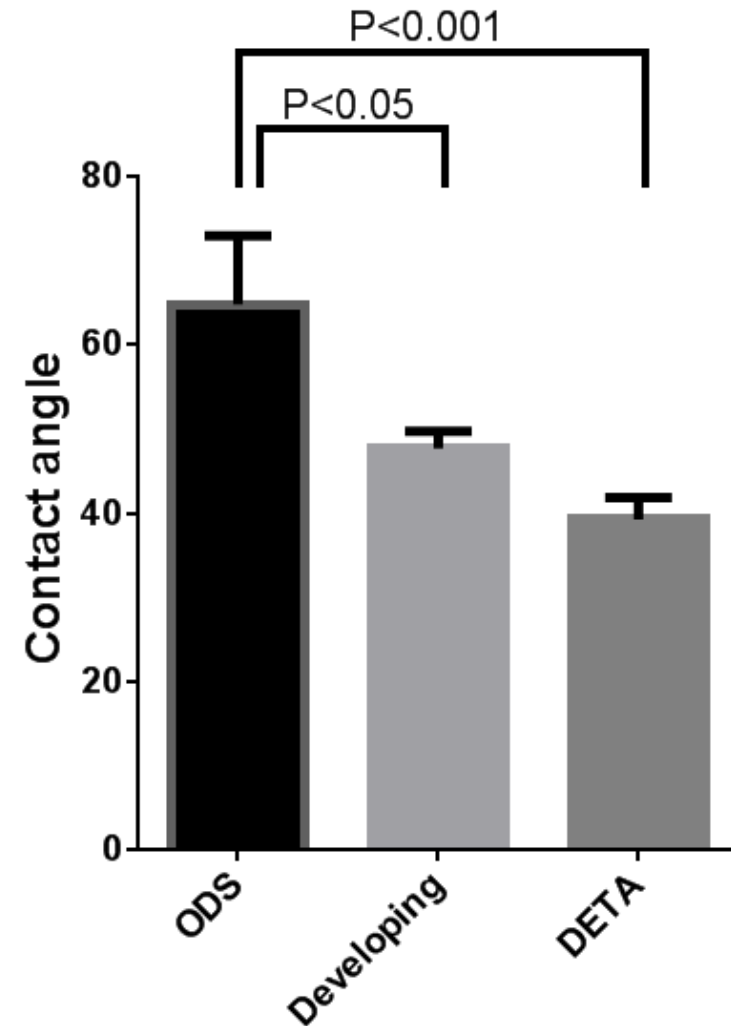
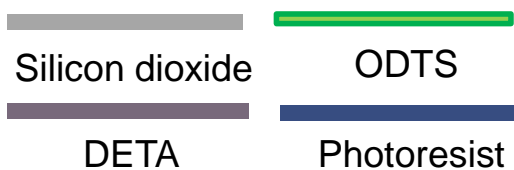
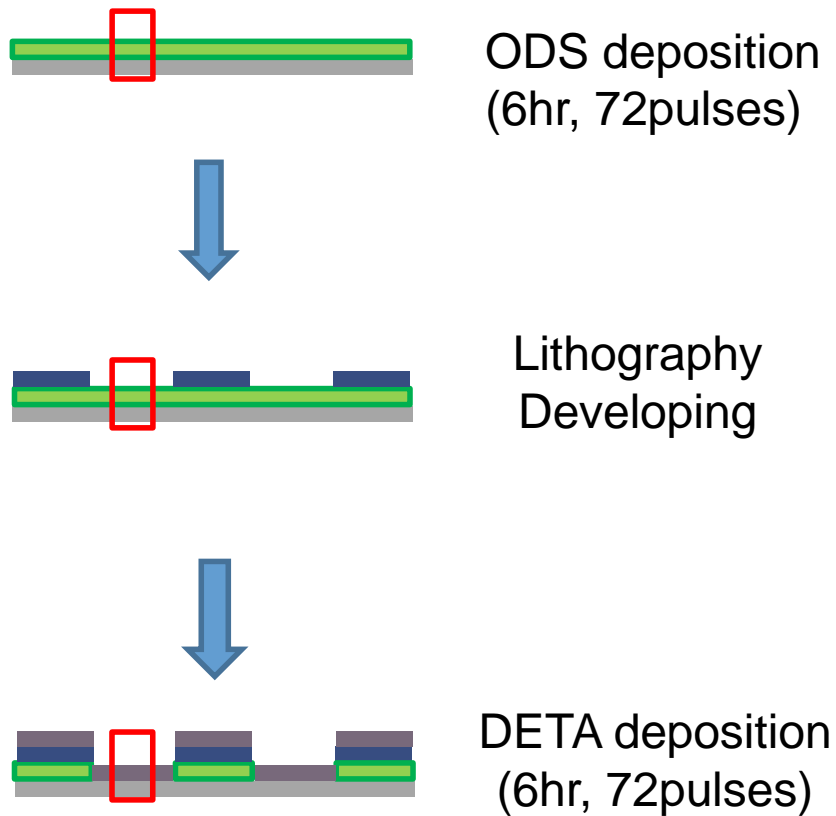
Isothiocyanate



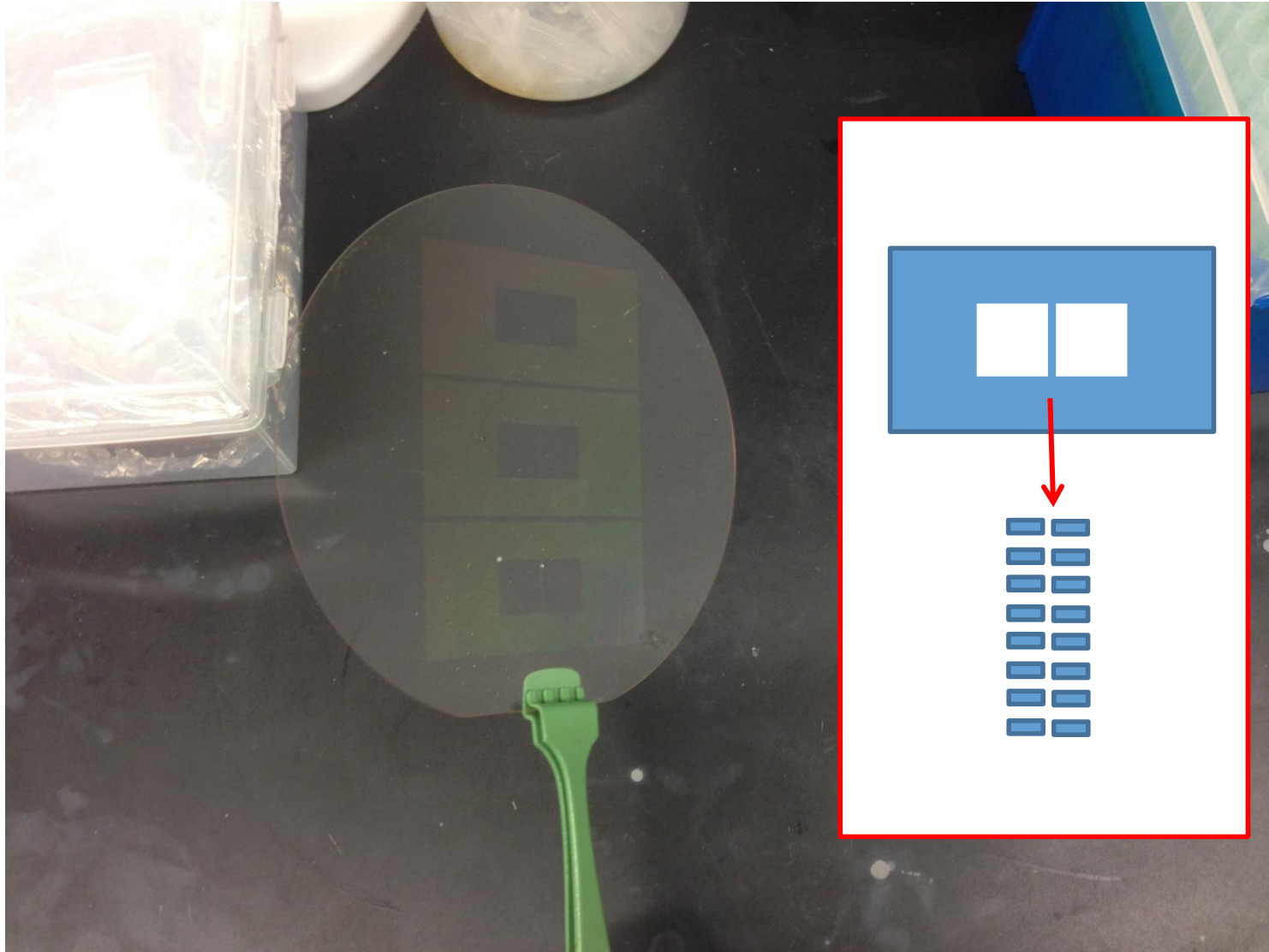
Characterization of DETA deposition



Fabrication of patterned ODS-DETA



Pattern of photoresist



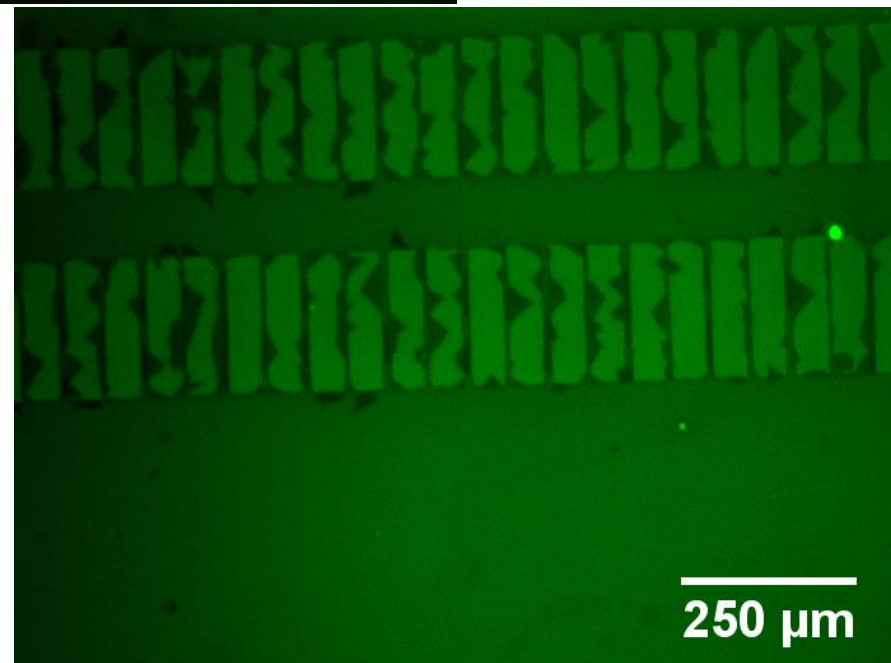
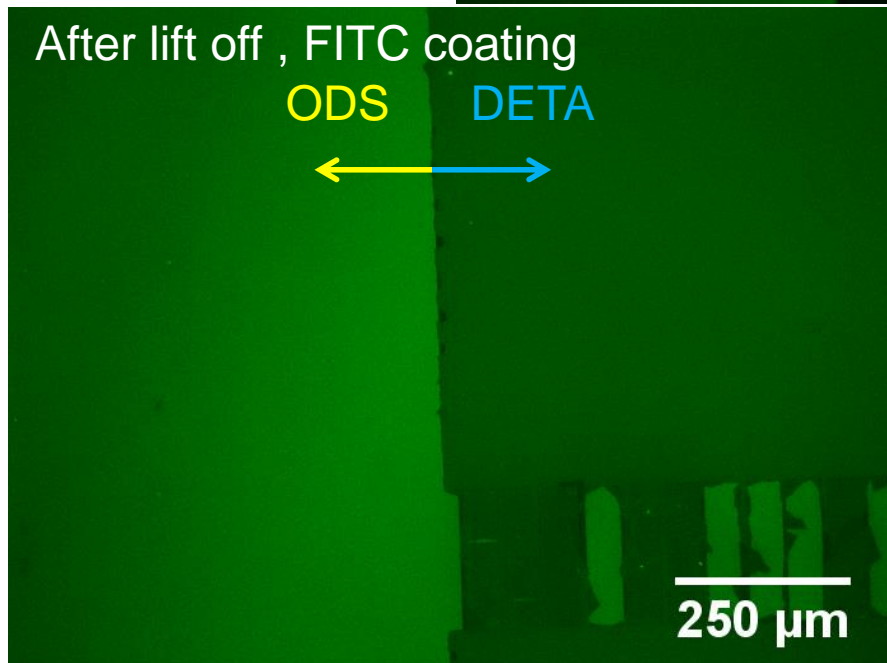
Before lift off , no FITC

Resist DETA

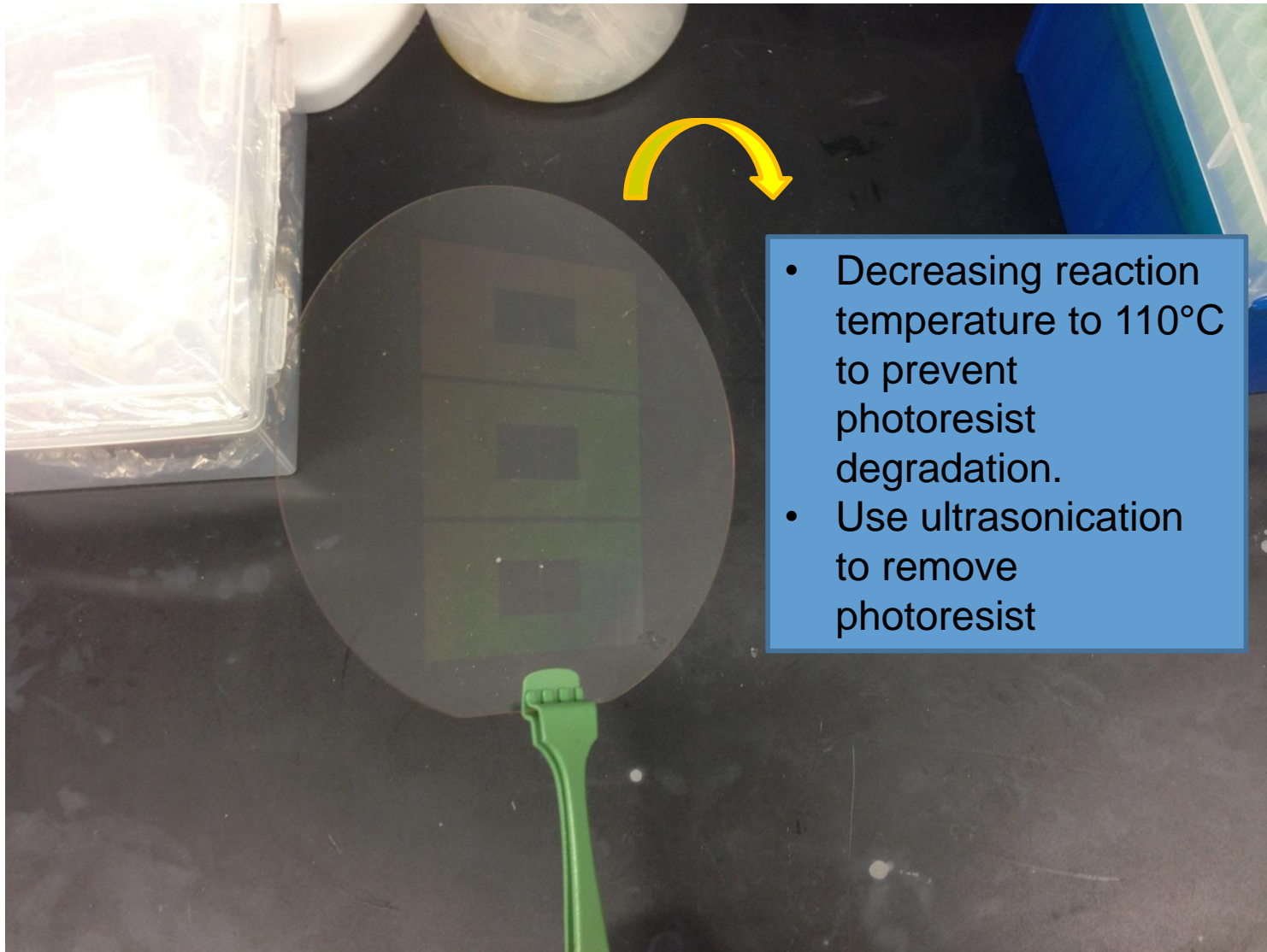


After lift off , FITC coating

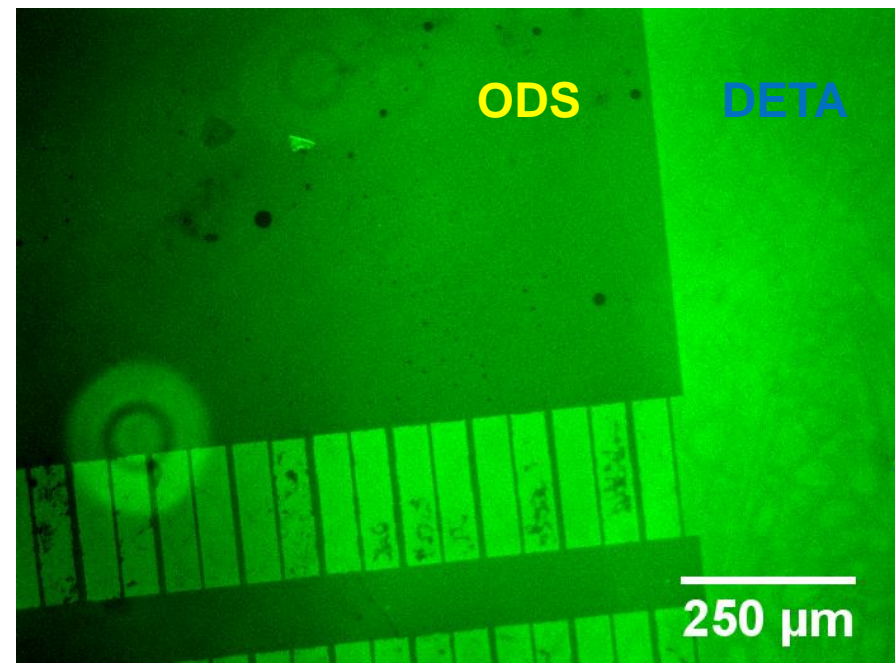
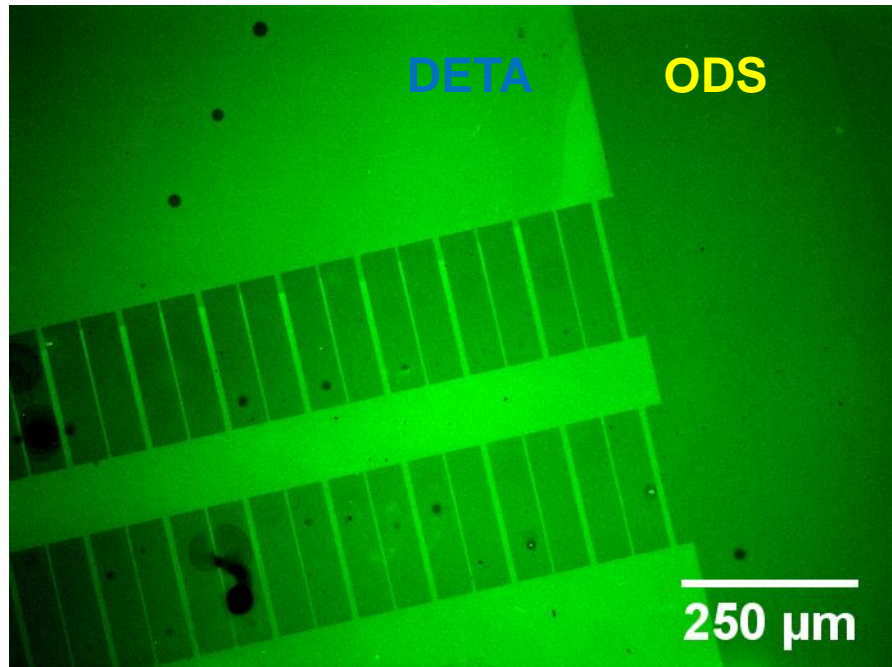
ODS DETA



Pattern of photoresist



- Decreasing reaction temperature to 110°C to prevent photoresist degradation.
- Use ultrasonication to remove photoresist



Summary

- Develop a recipe for ODS and DETA deposition.
- We used contact angle and Florescence spectroscopy to visualize the uniformity of the deposition.
- We applied photolithography to pattern a surface for neuron cells directed growth.
- Future plans is to grow neuron cells on the pattern template and apply the developed technique to modify electrodes for electrophysiology measurements.
- Optimization of ODS and DETA deposition, measuring the hydroxyl concentration on the surface. (FTIR)

Acknowledgement

- Stanford Nanofabrication Facility
 - Dr. Uli Thumser
 - Dr. Mary Tang

- Project Mentor
 - Dr. Michelle Rincon
 - Dr. J Provine

- Prof. Roger T. Howe
- Prof. Bianxiao Cui

